

# Synthesis and Nicotinic Acetylcholine Receptor Binding Properties of *exo*-2-(2'-Fluoro-5'-pyridinyl)-7-azabicyclo[2.2.1]heptane: A New Positron Emission Tomography Ligand for Nicotinic Receptors

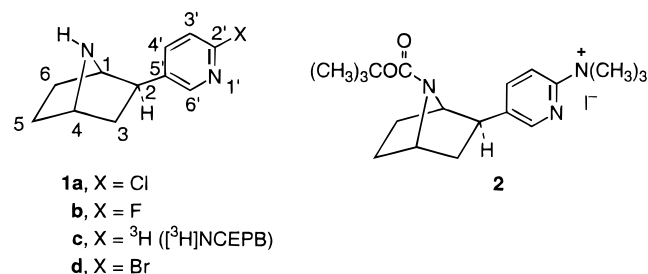
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Epibatidine, a unique alkaloid, was originally isolated from the skin of the Ecuadoran poison frog, *Epipedobates tricolor*, by Daly and co-workers and has been shown to have the structure *exo*-2-(2'-chloro-5'-pyridinyl)-7-azabicyclo[2.2.1]heptane (**1a**).<sup>1,2</sup> Later reports showed that the natural alkaloid possessed the 1*R*,2*R*,4*S* stereochemistry.<sup>3</sup> Numerous biological studies have demonstrated that epibatidine (**1a**) is a potent analgesic agent.<sup>4</sup> Its effects appear to be mediated via neuronal nicotinic receptors (nAChRs); however, details of this action at the molecular level have not been fully elucidated. Nevertheless, epibatidine (**1a**) has been a very useful tool for gaining new information concerning the pharmacological properties of neuronal nAChRs.<sup>4</sup>

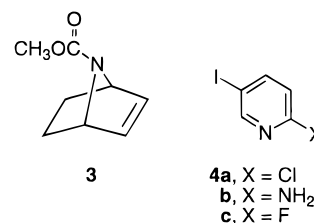
Recently, we synthesized and characterized the in vivo binding properties of (+)- and (-)-*exo*-2-(3-[6-<sup>3</sup>H]-pyridinyl)-7-azabicyclo[2.2.1]heptane (**1c**, [<sup>3</sup>H]norchloroepibatidine, [<sup>3</sup>H]NCEPB) in rats.<sup>5</sup> Both (+)- and (-)-[<sup>3</sup>H]NCEPB bind with high affinity but with higher levels of specific binding in vivo for nAChR as compared to epibatidine (**1a**).<sup>5</sup> Other investigators observed similar results with [<sup>3</sup>H]epibatidine.<sup>6</sup> These studies suggested that *exo*-2-(2'-fluoro-5'-pyridinyl)-7-azabicyclo[2.2.1]heptane (**1b**, norchlorofluoroepibatidine, NFEP) might also have favorable binding properties and that the fluorine-18 analog would be useful as a positron emission tomography (PET) ligand for further characterization of the nAChRs.



In this report, we present a synthesis of **1b** which can be adapted to provide a new synthesis of **1a** and

compare the nAChR binding properties of **1a**, **b** to that of nicotine. We also report the synthesis of 7-(*tert*-butyloxycarbonyl)-2-*exo*-[2'-(*N,N,N*-trimethylammonium)-5'-pyridinyl]-7-azabicyclo[2.2.1]heptane iodide (**2**), which is a more efficient precursor<sup>7,8</sup> for the synthesis of [<sup>18</sup>F]-**1b** (70% radiochemical yield)<sup>7</sup> than *exo*-2-(2'-bromo-5'-pyridinyl)-7-azabicyclo[2.2.1]heptane (**1d**) or *N*-protected analogs.<sup>9–11</sup>

A number of different chemical syntheses<sup>3,12–23</sup> of epibatidine (**1a**) have been reported, including one from our laboratory.<sup>15</sup> A review of these methods suggested that **1b** might best be prepared by appropriate modification of the synthesis of epibatidine (**1a**) reported by Clayton and Regan.<sup>13</sup> Their synthesis involved subjecting the olefin (**3**) to a reductive Heck arylation using 2-chloro-5-iodopyridine (**4a**) followed by removal of the *N*-methyloxycarbonyl protecting group with hydrogen bromide in acetic acid. This is the shortest synthesis of epibatidine (**1a**) thus far reported and gives only the desired *exo* product (**1a**).



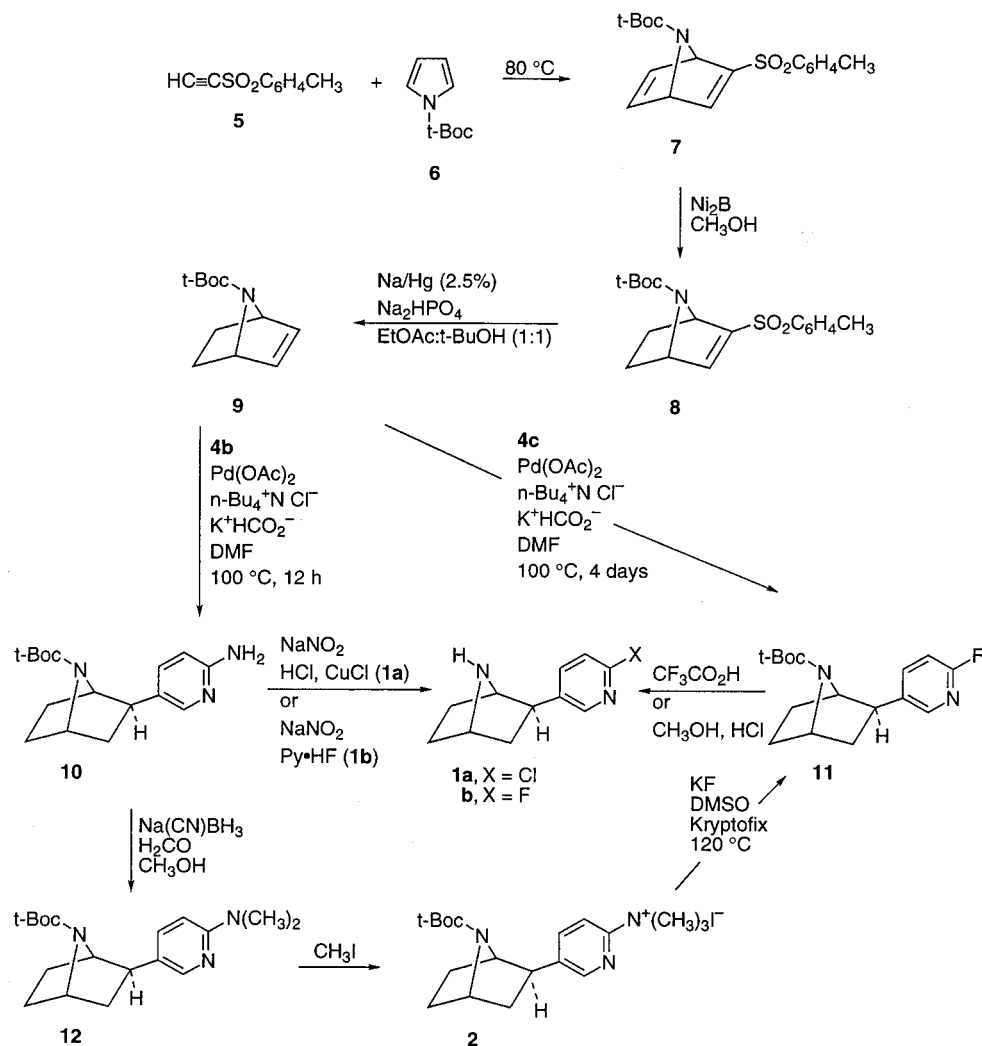
We found that **1b** could be prepared by the route shown in Scheme 1. Since we knew that the 2-fluoropyridinyl group of **1b** would be more reactive than the 2-chloropyridinyl group of epibatidine, we chose to use an *N*-*tert*-butyloxycarbonyl protecting group which could be removed under milder conditions in place of the *N*-methyloxycarbonyl group used by Clayton and Regan.<sup>13</sup> Thus, heating a solution of (*p*-tolylsulfonyl)acetylene<sup>13</sup> (**5**) and *N*-(*tert*-butoxycarbonyl)pyrrole<sup>24</sup> (**6**) provided 65% of the diene **7**. Nickel boride reduction of **7** gives 86% of the olefin **8**. Desulfonation of **8** using 2.5% sodium amalgam in a 1:1 mixture of ethyl acetate and *tert*-butyl alcohol yielded 55% of the desired 7-(*tert*-butoxycarbonyl)-7-azabicyclo[2.2.1]hept-2-ene (**9**). Coupling of **9** with 2-amino-5-iodopyridine (**4b**) using palladium acetate as catalyst in dimethylformamide containing tetrabutylammonium chloride and potassium formate provided 68% of 7-(*tert*-butoxycarbonyl)-*exo*-2-(2'-amino-5'-pyridinyl)-7-azabicyclo[2.2.1]heptane (**10**). The 2-*exo* stereochemistry assigned to **10** was based on an analysis of the <sup>1</sup>H NMR (CDCl<sub>3</sub>) spectrum. The spectrum showed a doublet of doublets at 2.74 ppm for the H-2 proton with  $J_{2\alpha,3\beta} = 5.0$  Hz,  $J_{2\alpha,3\alpha} = 8.8$  Hz, and  $J_{2\alpha,1} = 0$  Hz, which is characteristic of the 2-*exo* stereochemistry.<sup>15</sup> Diazotization<sup>25</sup> of **10** in pyridine containing 70% hydrogen fluoride effected conversion of the 2-amino group to a fluoro group and deprotection of the *N*-Boc group to give 46% of **1b**. Compound **1b** could also be obtained by direct reductive Heck coupling of 2-fluoro-5-iodopyridine (**4c**, obtained by diazotization of 2-amino-5-iodopyridine (**4b**)<sup>13</sup> in pyridine·HF) with **9** to give **11** which on removal of the *N*-Boc protecting group with trifluoroacetic acid gave the desired **1b**. However, this route was less desirable since the overall yield was 39% and the coupling of 2-fluoro-5-iodopyri-

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## Scheme 1



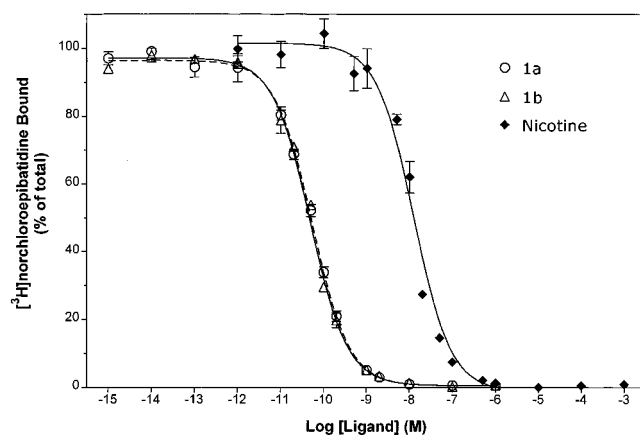
**Table 1.**  $K_{app}$  Values for **1a,b** and Nicotine at the nAChR<sup>a,b</sup>

compd	X	$K_{app}$ (nM) (SEM)
<b>1a</b>	Cl	0.024 ± 0.001
<b>1b</b>	F	0.020 ± 0.002
nicotine		6.2 ± 0.5

<sup>a</sup> The  $K_{app}$  values are reported as mean ± SEM and were calculated from 3–4 separate competition binding experiments. <sup>b</sup> Compound **1b** was characterized as its hydrochloride salt. C, H, and N analyses were within 0.4% of theoretical values.

dine (**4c**) with **9** took 4 days, whereas the coupling of 2-amino-5-iodopyridine (**4b**) with **9** was complete in 12 h. We also found that diazotization of **10** in hydrochloric acid in the presence or absence of cuprous chloride gave a 76% yield of epibatidine (**1a**) which was identical to an authentic sample.<sup>15</sup> This provided additional support for the 2-*exo* structural assignment of **1a,b**.

$K_{app}$  values for inhibition of [<sup>3</sup>H]NCEPB binding for **1a,b**, and nicotine were obtained using a modification of previously reported procedures.<sup>2,26</sup> The data are presented in Table 1 and Figure 1. For each compound, the IC<sub>50</sub> values obtained from three to four independent



**Figure 1.** [<sup>3</sup>H]Norchloroepibatidine competition binding experiments. The data represent the mean ± SEM from 3–4 binding experiments using 11–14 concentrations of test compound run in triplicate with different rat forebrain homogenate preparations for each experiment. The final assay concentration of [<sup>3</sup>H]norchloroepibatidine ( $K_d = 0.026 \pm 0.002$  nM;  $B_{max} = 5.7 \pm 0.05$  fmol/mg of tissue) was 30 pM.

competition binding experiments were used to calculate  $K_{app}$  values.<sup>27,28</sup> The 20 pM  $K_{app}$  value for **1b** makes this compound more than 100 times more potent than nicotine at the nAChR. A recent report has shown that [<sup>18</sup>F]-**1b** is an excellent PET ligand for mapping nAChR in vivo, and it provides the first evidence for using this

class of ligand to visualize nAChR in vivo with PET.<sup>7</sup> The studies demonstrated a high brain uptake of [<sup>18</sup>F]-**1b** in both baboon (12–15%) and mouse and high specificity of its binding for nAChR in vivo. The high thalamus to cerebellum ratio (4.0–4.5 in baboon at the end of a 2-h study) used as the index for specific to nonspecific binding provided a high signal-to-noise ratio suggesting a new approach to investigate the nAChR system and its role in neurodegeneration and addiction.

The synthesis of **2** and its conversion to **1b** are shown in Scheme 1. Reductive methylation of **10** using formaldehyde and sodium cyanoborohydride gives the 2-dimethylamino compounds **12**. Alkylation of **12** with methyl iodide affords 7-(*tert*-butyloxycarbonyl)-*exo*-2-[2'-(*N,N,N*-trimethylammoniumyl)-5'-pyridinyl]-7-azabicyclo[2.2.1]heptane iodide (**2**). Treatment of **2** with potassium fluoride in dimethyl sulfoxide containing Kryptofix gave a 75% yield of **11** which when treated with hydrochloric acid provided 55% of **1b**·HCl. Details for the synthesis of [<sup>18</sup>F]NFEP will be published elsewhere.

In summary, an efficient synthesis of *exo*-2-(2'-fluoro-5'-pyridinyl)-7-azabicyclo[2.2.1]heptane (**1b**) and epibatidine (**1a**) was developed. 7-(*tert*-Butyloxycarbonyl)-2-*exo*-[2'-(*N,N,N*-trimethylammoniumyl)-5'-pyridinyl]-7-azabicyclo[2.2.1]heptane (**2**) was synthesized as an excellent precursor for the synthesis of [<sup>18</sup>F]-**1b**.

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**Supporting Information Available:** Experimental data for the synthesis of **1a**, **2**, **7–11**, and **1b** from **10**, **11**, or **2** and the radioligand binding (7 pages). Ordering information is found on any current masthead page.

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- Under our experimental conditions, the concentration of free ligand was not much ( $>10\times$ ) greater than the number of receptors, a requirement if the binding data were to be analyzed as a pseudo-first-order bimolecular reaction. The result is the free ligand concentration and nonspecific binding will increase with increasing concentration of test compound leading to an underestimation of the  $K_i$ . However, for the purpose of this communication, a comparison of the  $K_{app}$  is sufficient to demonstrate the potency of **1b** relative to **1a** and nicotine. Thus, more sophisticated data analysis techniques designed to correct for changes in free ligand concentration were not employed.

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